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(54) Title: TALC CONTAINING POLYPROPYLENE COMPOSITIONS		
(57) Abstract <p>The present invention concerns polymer compositions and a process for the preparation thereof. The compositions comprise a propylene polymer nucleated with a polymeric nucleating agent containing vinyl compound units, and 0.1 to 10 % talc, calculated from the weight of the composition, to reduce shrinkage and to improve stiffness of the composition.</p>		

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TALC CONTAINING POLYPROPYLENE COMPOSITIONS

Background of the Invention**5 Field of the Invention**

The present invention relates to polymer compositions. In particular, the present invention concerns compositions of efficiently nucleated copolymers. The invention further concerns a process for preparing such compositions. Finally, the present invention relates to the use
10 of the new propylene polymers or copolymers for the manufacture of products by, e.g., blow moulding, thermoforming, injection moulding and sheet or film extrusion.

Description of Related Art

15 Propylene (PP) homo- and copolymers have excellent resistance to heat and chemicals as well as attractive mechanical properties, such as stiffness and impact resistance. However, processing of polypropylene by, e.g., injection moulding, thermoforming or blow moulding, to form thin-walled containers has resulted in products having insufficient transparency. This is caused by the semi-crystalline nature of polypropylene which leads to
20 hazy quality end-products.

In the prior art it has been proposed to improve the transparency of blow moulded polypropylene by blending the polymer with various nucleating agents such as dibenzilidene sorbitol (DBS) or di(alkylbenzilidene)sorbitol. These traditional nucleating
25 agents tend to bleed out from the polymer composition during processing and many of them give rise to fumes with an offensive smell. As a solution to these problems, it has been suggested in the art to use vinyl compounds, such as polymers of vinyl cycloalkanes and 3-methyl-1-butene, as nucleating agents in the form of propylene copolymers or polypropylene compounds, cf. EP Patent Specifications Nos. 0 152 701, 0 151 883, 0 368
30 577 and 0 417 319. However, studies done on these highly nucleated materials shows that, although their processing properties are excellent, shrinkage increases with increasing freezing point. Further, even though improved, the freezing point and stiffness of the

polymer resins are still too low for many applications.

Summary of the Invention

5 It is an object of the present invention to eliminate the problems related to the prior art and to provide a novel propylene polymer or copolymer composition having improved mechanical properties and therefore highly suited to the manufacture of moulded products.

A second object of the invention is to provide a novel process for preparing novel
10 propylene copolymer compositions of the above kind.

It is still a further object of the present invention to provide moulded or extruded products comprising propylene copolymer compositions.

15 These and other objects, together with the advantages thereof over known processes and products, which shall become apparent from the specification which follows, are accomplished by the invention as hereinafter described and claimed.

The invention is based on the idea of employing talc as a shrinkage-reducing ingredient in
20 propylene compositions, in particular in compositions containing propylene exhibiting a high degree of crystallinity and a high temperature of crystallization.

The use of talc in efficiently nucleated propylene compositions is known in the art. Thus, EP Patent Specification No. 0 152 701 discloses blending propylene copolymer containing
25 poly(vinyl cycloalkane) with a filler such as mica or talc. No examples of any kind of compositions containing talc are given nor does the specification give any concentration ranges for mica or talc. Considering the fact that talc is used as a filler, it would appear that amounts of several tens of percents have been contemplated.

30 Talc is also a well-known nucleating agent. EP Patent Application No. 0 586 109 describes a polypropylene composition containing 0.0001 to 1 part by weight of a crystal nucleating agent, such as talc. In the examples, the amount of talc is 0.2 parts by weight.

No indication of any beneficial effect of talc on the shrinkage of the polypropylene composition can be found in the above documents.

5 According to the present invention it has now been found that surprisingly good mechanical results are obtained by blending talc with propylene polymers which have been nucleated with a polymerized vinyl compound in amounts sufficient to obtain an increase of stiffness while reducing shrinkage to an acceptable level. Preferably 0.1 to 10 wt-% of talc is used. Surprisingly, although the stiffness of polypropylene nucleated with a polymerized vinyl compound is similar to that of polypropylene nucleated with 0.5 % talc,
10 the stiffness of the former will increase more than of the latter when both are blended with the same amount of talc (e.g. 2.5 to 5 %).

More specifically, the polymer composition according to the present invention is characterized by what is stated in the characterizing part of claim 1.

15

The process according to the present invention for preparing nucleated polypropylene compositions is characterized by what is stated in the characterizing part of claim 9.

20 The present compositions can be used in any kind of polymer articles. Particular advantages are obtained by applying the compositions to the manufacture of moulded products, and products prepared by blow moulding, stretch blow moulding, injection moulding, compression moulding and sheet or film extrusion and pipe or cable extrusion.

25 The invention achieves a number of considerable advantages. In particular it can be noted that the addition of talc in amounts of 0.1 to 10 wt-% increases the stiffness of the propylene polymer composition by up to 5 %. Talc in polypropylene compositions gives rise to higher tensile modulus than talc in standard PP copolymer. The Heat Deflection Temperature (the HDT value) is also increased by the addition of talk, and the HDT value increases more for the present polypropylene compositions nucleated with a vinyl
30 compound than for standard PP. The crystallization temperature of the present compositions is higher than for standard PP containing corresponding amounts of talc and for polypropylene compositions nucleated with a vinyl compound. Although the shrinkage

of the present compositions is somewhat higher than that of standard PP containing talc it is still on an acceptable level and the present invention provides a unique combination of excellent stiffness (up to 1600 MPa or more), controlled shrinkage, faster crystallization and high T_{cr} giving good cyclus potential. In the present context, faster crystallization is not necessarily related to faster formation of single crystals but rather to the fact that there are more crystals that fill the space faster.

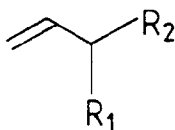
Next, the invention will be more closely examined with the aid of the following detailed description.

Detailed Description of the Invention

The compositions of the present invention include a polymeric component comprising a propylene polymer nucleated with vinyl compounds together with a shrinkage-restricting component, talc. By nucleating propylene polymers with vinyl compounds it is possible to provide polypropylene having a higher degree of crystallinity, a higher crystallization temperature, smaller crystallization size and a greater crystallization rate. These kinds of compositions can be used for the preparation of moulded products. They exhibit improved physical properties.

The nucleation of the propylene polymers can be carried out by modifying the polymerization catalyst with vinyl compounds and using the modified catalyst for polymerization of propylene optionally in the presence of comonomers to provide a propylene polymer or copolymer containing about 0.0001 to 1 % (calculated from the weight of the composition) of a polymer comprising vinyl compound units. Another approach for nucleating propylene polymers comprises blending polypropylene with polymers containing vinyl compound units.

For the purpose of the present invention "vinyl compounds" are compounds having the formula I



wherein R_1 and R_2 together form a 5 or 6 membered saturated or unsaturated or aromatic ring or they stand independently for a lower alkyl comprising 1 to 4 carbon atoms.

The following specific examples of vinyl compounds can be mentioned: vinyl
5 cycloalkanes, in particular vinyl cyclohexane, vinyl cyclopentane, vinyl-2-methyl
cyclohexane and vinyl norbornane, 3-methyl-1-butene, styrene, p-methyl-styrene, 3-ethyl-
1-hexene or mixtures thereof. Vinyl cyclohexane (VCH) is a particularly preferred
monomer but, for example 3-methyl-1-butene can be used as a monomer or comonomer to
adjust the crystallisation temperature.

10

For the purpose of the present invention "nucleated propylene polymer" stands for a
polymer having an increased and controlled degree of crystallinity and preferably having a
crystallization temperature of more than 7 °C, preferably over 10 °C and in particular over
13 °C of the T_{cr} of the corresponding non-nucleated polymer. Using $MgCl_2$ -supported high-
15 yield Ziegler-Natta catalysts crystallization temperatures of more than 120 °C, preferably
over 124 °C and in particular over 126 °C can be obtained. In compositions containing
colouring pigments having a nucleating effect, particularly advantageous results are
obtained by using polymers having a crystallization temperature over 15 °C higher than
that of the corresponding non-nucleated polymer (for a polymer produced with the above-
20 mentioned ZN-catalyst, 128 °C).

According to a preferred embodiment of the present invention, modification of the catalyst
with by polymerizing a vinyl compound, such as VCH, in the presence thereof is
performed in an inert fluid which does not dissolve the polymer formed (e.g. polyVCH).

25

One particularly preferred polymerization medium comprises a viscous substance, in the
following a "wax", such as an oil or a mixture of an oil with a solid or semi-solid polymer
(oil-grease). The viscosity of such a viscous substance is usually 1,000 to 15,000 cP at
room temperature. The advantage of wax prepolymerization is that the catalyst can be
prepolymerized, stored and fed into the process in the same media and catalyst wax
30 preparation and prepolymerization is performed in the same process device. Due to the
fact that no washing, drying, sieving and transferring thus are needed, the catalyst activity
is maintained (cf. Finnish Patent No. 95387). The present process is inexpensive because

high catalyst concentrations and high PP production capacities can be used. Also the amount of waste is diminished because the medium used during polymerization of the vinyl compound does not have to be removed.

5 According to another preferred embodiment, the method for improving the crystallinity and transparency of polypropylene by blending a crystalline polypropylene with a vinyl cycloalkane polymer is carried out by melt-kneading the crystalline polypropylene with the crystal nucleating agent, compounding the crystal nucleating agent with the crystalline polypropylene and melt kneading the mixture during film formation, and compounding the
10 master batch of the crystal nucleating agent with the crystalline polypropylene.

The vinyl compound units of the blending and compounding process can be derived from any of the units identified in the above formula I in connection with the first embodiment of the invention.

15

As catalyst any stereospecific catalyst for propylene polymerization can be used, which is capable of catalyzing polymerization and copolymerization of propylene and comonomers at a pressure of 10 to 100 bar, in particular 25 to 80 bar, and at a temperature of 40 to 110 °C, in particular 60 to 100 °C. Ziegler-Natta as well as metallocene catalysts can be used.

20

Generally, the Ziegler-Natta catalyst used in the present invention comprises a catalyst component, a cocatalyst component, an external donor, the catalyst component of the catalyst system primarily containing magnesium, titanium, halogen and an internal donor.

25

Examples of suitable catalyst systems are described in, for example, Finnish Patents Nos. 86866, 96615 and 88047 and 88048.

30

One particularly preferable catalyst, which can be used in the present invention, is disclosed in FI Patent No. 88047. Another preferred catalyst is disclosed in Finnish Patent Application No. 963707.

A catalyst system useful in the present process can be prepared by reacting a magnesium

halide compound with titanium tetrachloride and an internal donor. The magnesium halide compound is, for example, selected from the group of magnesium chloride, a complex of magnesium chloride with a lower alkanol and other derivatives of magnesium chloride. MgCl₂ can be used as such or it can be combined with silica, e.g. by absorbing the silica
5 with a solution or slurry containing MgCl₂. The lower alkanol used can be preferably methanol or ethanol, particularly ethanol.

The titanium compound used in the preparation of the procatalyst is preferably an organic or inorganic titanium compound, having an oxidation state of titanium of 3 or 4. Also other
10 transition metal compounds, such as vanadium, zirconium, chromium, molybdenum and tungsten compounds can be mixed with the titanium compound. The titanium compound usually is halide or oxyhalide, an organic metal halide, or a purely metal organic compound, in which only organic ligands have been attached to the transition metal. Particularly preferable are the titanium halides, especially TiCl₄. Preferably the titaniation is
15 carried out in two or three steps.

The Ziegler-Natta catalyst used can also be an heterogeneous unsupported TiCl₃ based catalyst. This kind of catalysts are typically solid TiCl₃ in a delta crystalline form which are activated with aluminium-chloride-alkyls, such as diethylaluminiumchloride. The solid
20 TiCl₃ catalyst are typically prepared by reduction of TiCl₄ with aluminium-alkyls and/or aluminium-chloride-alkyls, possibly combined with heat treatment to maximise the desired delta crystalline form of TiCl₃. The performance, especially stereospecificity, of these catalyst can be improved by using Lewis-bases (electron donors), such as esters, ethers or amines.

25

The activity and stereospecificity of these heterogeneous unsupported TiCl₃ based catalysts are typically so low that propylene polymers or copolymers prepared require purification from catalyst residues and removal of some atactic non-crystalline polymer.

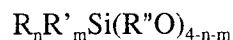
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The alkoxy group of the phthalic acid ester used comprises at least five carbon atoms, preferably at least 8 carbon atoms. Thus, as the ester can be used for example propylhexyl phthalate, dioctyl phthalate, dinonyl phthalate, diisodecyl phthalate, di-undecyl phthalate,

ditridecyl phthalate or ditetradecyl phthalate.

The partial or complete transesterification of the phthalic acid ester can be carried out e.g. by selecting a phthalic acid ester - a lower alcohol pair, which spontaneously or with the aid of a catalyst, which does not damage the procatalyst composition, transesterifies the catalyst at an elevated temperatures. It is preferable to carry out the transesterification at a temperature, which lies in the range of 110 to 150 °C, preferably 120 to 140 °C.

The catalyst prepared by the method above is used together with an organometallic cocatalyst and with an external donor. Generally, the external donor has the formula IV



wherein

- 15 R and R' can be the same or different and they stand for a linear, branched or cyclic aliphatic, or aromatic group;
- R'' is methyl or ethyl;
- n is an integer 0 to 3;
- m is an integer 0 to 3; and
- 20 n+m is 1 to 3.

The aliphatic groups in the meanings of R and R' can be saturated or unsaturated. Linear C₁ to C₁₂ hydrocarbons include methyl, ethyl, propyl, butyl, octyl and decanyl. As examples of suitable saturated branched C₁₋₈ alkyl groups, the following can be mentioned: isopropyl, isobutyl, isopentyl, tert-butyl, tert-amyl and neopentyl. Cyclic aliphatic groups containing 4 to 8 carbon atoms comprise, e.g., cyclopentyl, cyclohexyl, methyl cyclopentyl and cycloheptyl.

A group of interesting donors is formed by strongly co-ordinating donors which form relatively strong complexes with catalyst surface, mainly with MgCl₂ surface in the presence of aluminium alkyl and TiCl₄. The donor components are characterised by a strong complexation affinity towards catalyst surface and a sterically large and protective

hydrocarbon (R'). Such external donors can be selected from the group consisting of dicyclopentyl dimethoxysilane, diisopropyl dimethoxysilane, di-isobutyl dimethoxysilane, and di-t-butyl dimethoxysilane.

- 5 An organoaluminum compound is used as a cocatalyst. The organoaluminium compound is preferably selected from the group consisting of trialkylaluminium, dialkyl aluminium chloride and alkyl aluminium sesquichloride.

The metallocene catalyst comprises a metallocene/activator reaction product impregnated
10 in a porous support at maximum internal pore volume. The catalyst complex comprises a ligand which is typically bridged, and a transition metal of group IVA...VIA, typically a metal halide, and aluminium alkyl. The ligands can belong to group of heterocyclic substituted or unsubstituted compounds, e.g. indocenes, naftenes, or any other bulky compound which can control the stereoselectivity of the catalyst especially when ligands
15 are bridged together with silane or other chemical bond. The activator is selected from a group in which are derivatives of water and aluminium alkyls e.g. trimethyl aluminium, triethyl aluminium, and tri t-butyl aluminium, or another compound capable of activating the complex. The metallocene/activator reaction product, a solvent capable of dissolving it, and a porous support are brought into mutual contact, the solvent is removed and the
20 porous support is impregnated with the metallocene/activator reaction product, the maximum amount of which corresponds to the pore volume support, cf. International PCT Application No. PCT/FI94/00499.

One typical structure of metallocene compound reactions is bridged bis(2-R-4-R'-indenyl)
25 $M Cl_2$, wherein both R and R' are aliphatic, cycloaliphatic or aromatic hydrocarbons having 1 to 18 C atoms, R' is typically phenyl or naphthyl, and R is typically methyl or ethyl. M is a transition metal, typically titanium, zirconium or hafnium. R and R' may contain heteroatoms, such as silicon, nitrogen, phosphorous or germanium. The bridge between the indenyls is made from 1 to 3 atoms, such as carbon, silicon, nitrogen,
30 phosphorous or germanium. A typical bridge is dimethylsilyl or ethyl. An example of such a metallocene compound is dimethylsilyl-bis(2-methyl-4-phenyl-indenyl)zirconium dichloride.

Following the modification of the catalyst with the vinyl compound of the first preferred embodiment of the invention, the catalyst is optionally prepolymerized with propylene and/or another 1-olefin to provide a prepolymerized catalyst composition which is used for polymerization of propylene optionally together with comonomers.

5

The propylene homo- or copolymer can have a unimodal or bimodal molar mass distribution. Thus, the equipment of the polymerization process can comprise any polymerization reactors of conventional design for producing propylene homo- or copolymers. For the purpose of the present invention, "slurry reactor" designates any reactor, such as a continuous or simple batch stirred tank reactor or loop reactor, operating in bulk or slurry and in which the polymer forms in particulate form. "Bulk" means a polymerization in reaction medium that comprises at least 60 wt-% monomer. According to a preferred embodiment the slurry reactor comprises a bulk loop reactor. By "gas phase reactor" is meant any mechanically mixed or fluid bed reactor. Preferably the gas phase reactor comprises a mechanically agitated fluid bed reactor with gas velocities of at least 0.2 m/sec.

Thus, the polymerization reactor system can comprise one or more conventional stirred-tank slurry reactors, as described in WO 94/26794, or one or more gas phase reactors. Preferably the reactors used are selected from the group of loop and gas phase reactors and, in particular, the process employs at least one loop reactor and at least one gas phase reactor. This alternative is particularly suitable for producing bimodal polypropylene. By carrying out the polymerization in the different polymerization reactors in the presence of different amounts of hydrogen, the MWD of the product can be broadened and its mechanical properties improved. It is also possible to use several reactors of each type, e.g. one loop reactor and two or three gas phase reactors or two loop reactors and one gas phase reactor, in series.

In every polymerization step it is possible to use also comonomers selected from the group of ethylene, propylene, butene, pentene, hexene and alike as well as their mixtures.

30

In addition to the actual polymerization reactors used for producing the propylene homo-

or copolymer, the polymerization reaction system can also include a number of additional reactors, such as pre- and/or postreactors. The prereactors include any reactor for prepolymerizing the catalyst with propylene and/or other 1-olefins. The postreactors include reactors used for modifying and improving the properties of the polymer product.

5 All reactors of the reactor system are preferably arranged in series.

The gas phase reactor can be an ordinary fluidized bed reactor, although other types of gas phase reactors can be used. In a fluidized bed reactor, the bed consists of the formed and growing polymer particles as well as still active catalyst come along with the polymer

10 fraction. The bed is kept in a fluidized state by introducing gaseous components, for instance monomer on such flowing rate which will make the particles act as a fluid. The fluidizing gas can contain also inert carrier gases, like nitrogen and also hydrogen as a modifier. The fluidized gas phase reactor can be equipped with a mechanical mixer.

15 The gas phase reactor used can be operated in the temperature range of 50 to 115 °C, preferably between 60 and 110°C and the reaction pressure between 5 and 50 bar and the partial pressure of monomer between 2 and 45 bar.

The pressure of the effluent, i.e. the polymerization product including the gaseous

20 reaction medium, can be released after the gas phase reactor in order optionally to separate part of the gaseous and possible volatile components of the product, e.g. in a flash tank. The overhead stream or part of it is recirculated to the reactor.

The propylene homo- or copolymer produced preferably has a MWD of over 2 to 10 and

25 a MFR₂ in the range of 0.01 to 1000 g/10 min, preferably 0.05 to 500 g/10 min, measured by ISO 1133, 230 °C, 2.16 kg load.

In the second embodiment of the invention, wherein a uni- or bimodal propylene homo- or copolymer is blended and compounded with a polymer comprising vinyl compound

30 units, the blending is carried out as known in the art using said nucleating polymeric agent.

By means of both embodiments, a propylene homopolymer or copolymer is produced having high stiffness, an increased overall degree of crystallization and a crystallization temperature measured with DSC of more than 7 °C, preferably over 10 °C and in particular 13 °C higher than that of the corresponding non-nucleated polymer. The degree of crystallization for the propylene homopolymer is generally over 48 %, often over 50 %, and the elasticity modulus is about 2,000 MPa or more. The elasticity modulus of heterophasic copolymers containing about 12 wt-% of a rubbery component is about 1,500 MPa or more.

If desired, the polymerization product can be fed into a gas phase reactor in which a rubbery copolymer is provided by a (co)polymerization reaction to produce a modified polymerization product. This polymerization reaction will give the polymerization product properties of e.g. improved impact strength. The step of providing an elastomer can be performed in various ways. Thus, preferably an elastomer is produced by copolymerizing at least propylene and ethylene into an elastomer. The conditions for the copolymerization are within the limits of conventional EPM production conditions such as they are disclosed, e.g., in Encyclopedia of Polymer Science and Engineering, Second Edition, Vol. 6, p.545-558. A rubbery product is formed if the ethylene repeating unit content in the polymer lies within a certain range. Thus, preferably, ethylene and propylene are copolymerized into an elastomer in such a ratio that the copolymer contains from 10 to 70 % by weight of ethylene units. In particular, the ethylene unit content is from 30 to 50 % by weight of the amorphous copolymer propylene/ethylene elastomer. In other words, ethylene and propylene are copolymerized into an elastomer in a molar ratio of ethylene-to-propylene of 30/70 to 50/50. Polymers modified by adding the rubbery copolymer in a gas phase reactor are typically called polypropylene block copolymers or heterophasic copolymers.

The amount of the nucleating polymeric agent is, in case of propylene copolymers, about 0.0001 to 1 wt-%, and in case of polypropylene blends about 0.0001 to 0.5 wt-%.

This nucleated polypropylene is then blended with talc in an amount of 0.1 to 10 wt-%, preferably about 0.5 to 9 wt-%, in particular 1 to 8 wt-% (calculated from the total weight of the composition). The talc used has a particle size in the range of 0.01 to

100 μm , preferably 0.1 to 10 μm .

The present blends can be produced by methods known *per se*, e.g. by mixing the polymer components with the talc in the desired weight relationship using a batch or a continuous
5 process. As examples of typical batch mixers, the Banbury and the heated roll mill can be mentioned. Continuous mixers are exemplified by the Farrel mixer, the Buss co-kneader, and single- or twin-screw extruders.

A composition according to the invention containing some 1 to 8 wt-% talc exhibits at
10 least about 10 %, preferably about 12 % to about 25 %, lesser shrinkage than the corresponding composition not containing talc. This decreased shrinkage is manifested particularly transverse to flow direction. Too high talc loading will impair a further increase of stiffness and restrict shrinkage below the lower tolerance of the moulds conventionally used for processing in the melt stage.

15 As mentioned above, the mechanical properties of polymer compositions according to invention are generally good to excellent. Thus, with a loading of 1 to 8 wt-% talc, a polymer composition of a heterophasic propylene copolymer containing about 12 % of a rubbery component will have a stiffness (tensile modulus) of over 1500 MPa and a HDT of
20 over 106 °C. Generally, a loading of 2.5 % talc will increase the tensile modulus from, e.g., 1440 MPa to 1600 MPa and a loading of 5 % will further increase the tensile modulus to 1750 MPa (tensile modulus measured by using ISO 527-2). Impact properties are not influenced by the addition of talc in the range indicated, in particular if the amount of talc is less than 5 %.

25 The homopolymer or copolymer composition thus obtained can be used for the manufacture of moulded articles, in particular articles processed by blow moulding, thermoforming, injection moulding and sheet or film extrusion, or pipe or cable extrusion. The compositions of the invention are particularly suitable for pipes and fitting for non-
30 pressure sewage or pressure application, and buffer tubes, e.g. for optical fibres, wherein the amount of polymerized vinyl compound in the propylene polymer is less than 0.01 wt-%.

Examples

The following non-limiting examples illustrate the invention.

5

Example 1

Catalyst modification with polymerisation of vinylcycloalkane:

A high yield $MgCl_2$ supported $TiCl_4$ Ziegler-Natta catalyst prepared according to Finnish
10 patent FI 88047 was dispersed into a mixture of oil and grease (Shell Ondina Oil N 68 and
Fuchs Vaseline Grease SW in 3.2:1 oil/grease volume ratio). The titanium content of the
catalyst was 2.5 wt-%, and the concentration of the catalyst in the oil/grease mixture was
189 g cat/dm³. Triethylaluminium (TEAL) was added to the catalyst dispersion in a TEAL
to titanium mole ratio of 1.5. After that vinylcyclohexane (VCH) was added to the reaction
15 mixture, and the VCH to catalyst weight ratio was 1:1. The reaction mixture was mixed at
a temperature of 55 °C until the concentration of unreacted VCH in the reaction mixture
was 1000 ppm by weight.

Example 2

20 Copolymerisation of propylene and ethylene

The modified catalyst in the oil-grease mixture (catalyst mud) obtained from Example 1
was continuously fed to process consisting from a loop reactor and a fluid bed gas phase
reactor together with TEAL, dicyclopentyl dimethoxysilane and propylene.

25

The TEAL and dicyclopentyl dimethoxysilane at a weight-to-weight ratio of 3.2 was
contacted before mixing with the catalyst mud. After that the mixture was flushed with
propylene, containing the desired amount of hydrogen as molecular weight regulating
agent, to a continuous stirred prepolymerisation reactor. After the prepolymerisation, the
30 reaction mixture together with additional propylene and hydrogen was fed to a continuous
loop reactor operating at 68 °C. The resulted PP homopolymer-propylene slurry containing
the catalyst was continuously recovered from the loop reactor to a flashing unit where the

liquid propylene was vaporised and the remaining solid polymer particles, containing the active modified catalyst, was further fed to a continuous fluidised bed gas phase reactor where a rubbery propylene ethylene copolymer for impact modification was produced. The gas phase reactor was operated at 77 °C. The desired amount of propylene and ethylene was continuously fed to the reactor, and the molecular weight of the copolymer produced was controlled with desired amount of hydrogen. The final polymer was continuously recovered from the gas phase reactor. After purging the unreacted monomers, the required stabilisers and other additives were added and the polymer powder was pelletised with an extruder.

10

The final polymer had an MFR₂ of 15 g/10 min, and an ethylene content of 8 wt-%. The amount of polymer precipitated with acetone (AM) from solubles in 23 °C xylene (XS) was 13.9 wt-%, which corresponds to the amount of rubbery copolymer made in the gas phase reactor. The MFR₂ of the polymer recovered from the loop reactor was 20 g/10 min.

15

The ethylene content of the polymer precipitated with acetone from solubles in 23 °C xylene was 37.4 wt-%. The polymer solubles in xylene at 25 °C were measured by dissolving the polymer in xylene at 135 °C, and allowing the solution to cool to 25 °C and filtering then the insoluble part out. The amorphous part, AM, was measured by separating the above xylene soluble fraction and precipitating the amorphous part with acetone.

20

Example 3

Blending of nucleated PP with talc

25

The nucleated polypropylene prepared according to Example 2 was blended with talc and compounded in a extruder. The stiffness of the present compositions were compared with those of a standard nucleated PP copolymer. The nucleation agent of the standard copolymer was talc (<0.5 %) or Sodium Benzoate (<0.1 %).

30

Tensile modulus were measured according to ISO 527-2:

			Shrinkage
	Standard PP polymer	Tensile modulus 1,400MPa	1.47 %
	Standard PP copolymer + 3 % talc	Tensile modulus 1,420 MPa	1.38 %
	Copolymer of Example 2	Tensile modulus 1,400 MPa	1.70 %
5	Copolymer of Example 2 +3 % talc	Tensile modulus 1,600 MPa	1.59 %

Shrinkage is measured from open top containers according to Borealis "Injection moulding application method 4".

10 The results show that by blending a copolymer nucleated with polymerized vinyl compounds and talc a surprising increase in the tensile modulus can be obtained; the propylene copolymer of Example 2 responded to talc with much higher modulus than the standard PP copolymer. The shrinkage values show that the invention provides a unique combination of excellent stiffness and controlled shrinkage.

15

Example 4

The effect on HDT and Tc

20 The nucleated polypropylene prepared according to Example was blended with talc and compounded in a extruder. The stiffness of the present compositions were compared with those of a standard nucleated PP copolymer. The nucleation agent of the standard copolymer was talc (<0.5 %) or Sodium Benzoate (<0.1 %).

25 Heat Deflection Temperature (HDT) was measured according to ISO- 75-2 (method B/0.45 MPa).

Temperature of crystallisation was measured with differential scanning calorimeter (DSC) at a cooling rate of 10 °C/min.

30	Standard PP copolymer	HDT 100 °C	T _{cr} 123 °C
	Standard PP copolymer +3 % talc	HDT 104 °C	T _{cr} 127 °C
	Copolymer of Example 2	HDT 103 °C	T _{cr} 129 °C

Copolymer of Example 2 +3 % talc HDT 109 °C T_{cr} 131 °C

5 The above results clearly show that, when blended with talc, polymers nucleated with polymerized vinyl compounds have better properties than the pure nucleated PP polymer and the standard PP copolymer blended 3 % talc.

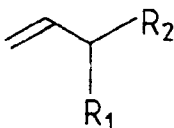
Thus, the above results indicate that a synergistic effect on heat deflection temperature (HDT) and temperature of crystallisation (T_{cr}) can be obtained by blending polymerically nucleated PP and talc.

IN THE CLAIMS:

1. Polymer composition comprising

- 5 – a propylene polymer nucleated with a polymeric nucleating agent containing vinyl compound units, and
- 0.1 to 10 % talc, calculated from the weight of the composition, to reduce shrinkage and to improve stiffness of the composition.

- 10 2. The polymer composition according to claim 1, wherein the propylene polymer contains about 0.0001 to 1 % by weight of a vinyl compound of the formula



- 15 wherein R₁ and R₂ together form a 5 or 6 membered saturated or unsaturated or aromatic ring or they stand independently for a lower alkyl comprising 1 to 4 carbon atoms.

- 20 3. The polymer composition according to claim 2, wherein the propylene polymer contains cycloalkane units, in particular vinyl cyclohexane, vinyl cyclopentane, vinyl-2-methyl cyclohexane and vinyl norbornane, 3-methyl-1-butene, styrene, p-methyl-styrene or 3-ethyl-1-hexene units or mixtures thereof.

- 25 4. The polymer composition according to any of claims 1 to 3, wherein the nucleated propylene polymer comprises a propylene copolymer containing vinyl compound units.

5. The polymer composition according to any of claims 1 to 3, wherein the nucleated propylene polymer comprises a propylene homo- or copolymer blended with a polymeric nucleating agent containing vinyl compound units.

- 30 6. The polymer composition according to any of claims 1 to 5, wherein the nucleated propylene homopolymer component of the composition exhibits a Melt Flow Rate (MFR₂) of 0.1 to 100 g/10 min, a T_c of over 7 °C higher than the T_c of the corresponding non-

nucleated polymer, a crystallinity of over 48 % and a elasticity modulus of over 2000 MPa.

7. The polymer composition according to any of claims 1 to 6, comprising 1 to 8 wt-% talc and exhibiting a 10 %, preferably 12 % to 20 %, lesser shrinkage than the
5 corresponding composition not containing talc.

8. The polymer composition according to any of claims 1 to 7, wherein the nucleated propylene heterophasic copolymer component of the composition exhibits a stiffness (tensile modulus) of over 1500 MPa and a HDT of over 106 °C.
10

9. A process for preparing a polymer composition comprising a propylene polymer nucleated with a polymeric nucleating agent containing vinyl compound units and exhibiting reduced shrinkage and improved stiffness, characterized by blending the propylene polymer with 0.1 to 10 % talc, calculated from the weight of the composition.
15

10. The process according to claim 9, wherein the propylene polymer is blended with 0.5 to 9 wt-%, preferably 1 to 8 wt-% of talc.

11. The process according to claim 9 or 10, wherein talc is used having a particle size
20 distribution of 0.1...10 µm.

12. The process according to any of claims 9 to 11, wherein 1 to 8 wt-% talc is blended with the propylene polymer nucleated with a polymeric nucleating agent containing vinyl compound units to provide a composition exhibiting a 10 %, preferably 12 % to 20 %, lesser shrinkage than the corresponding composition not containing talc.
25

13. Use of a polymer composition according to any of claims 1 to 8 for the manufacture of polymer articles by blow moulding, thermoforming, injection moulding, compression moulding and sheet or film extrusion or pipe or cable extrusion.
30

14. The use according to claim 13, wherein the polymer composition is used for manufacture of pipes and fittings for non-pressure sewage or pressure applications.

15. The use according to claim 13, wherein the polymer composition is used for manufacture of buffer tubes for optical fibre cables, said polymer composition containing polymerized vinyl compounds less than 0.01 wt-%.

INTERNATIONAL SEARCH REPORT

International Application No
PCT/EP 98/07216

A. CLASSIFICATION OF SUBJECT MATTER
IPC 6 C08K3/34 C08L53/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 6 C08K C08L

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 152 701 A (SUMITOMO CHEMICAL CO) 28 August 1985 cited in the application see page 8, line 1 see claim 1	1-15
A	---	
A	EP 0 265 075 A (MITSUI TOATSU CHEMICALS ;HONDA MOTOR CO LTD (JP)) 27 April 1988 see page 4, line 47 - line 48 see claim 1	1-15

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Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

° Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- "&" document member of the same patent family

Date of the actual completion of the international search

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INTERNATIONAL SEARCH REPORT

International Application No
 PCT/EP 98/07216

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	DATABASE WPI Section Ch, Week 8939 Derwent Publications Ltd., London, GB; Class A17, AN 89-281433 XP002098803 & JP 01 204947 A (NISSAN MOTOR CO LTD) , 17 August 1989 see abstract <div style="text-align: center; margin-top: 10px;">-----</div>	1-15

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP 98/07216

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 0152701 A	28-08-1985	JP 1782527 C	13-08-1993
		JP 3074247 B	26-11-1991
		JP 60139710 A	24-07-1985
		JP 1761591 C	28-05-1993
		JP 3042298 B	26-06-1991
		JP 60139731 A	24-07-1985
		CA 1255048 A	30-05-1989
		DE 3471942 A	14-07-1988
		US 4696979 A	29-09-1987
		CA 1254690 A	23-05-1989
		EP 0151883 A	21-08-1985
		SG 96994 G	12-05-1995
		US 4551501 A	05-11-1985
EP 0265075 A	27-04-1988	JP 1716143 C	27-11-1992
		JP 4002620 B	20-01-1992
		JP 63095252 A	26-04-1988
		CA 1300778 A	12-05-1992
		CN 1009103 B	08-08-1990
		DE 3774532 A	19-12-1991

INTERNATIONAL SEARCH REPORT

information on patent family members

International Application No

PCT/EP 98/07216

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 0152701 A	28-08-1985	JP 1782527 C	13-08-1993
		JP 3074247 B	26-11-1991
		JP 60139710 A	24-07-1985
		JP 1761591 C	28-05-1993
		JP 3042298 B	26-06-1991
		JP 60139731 A	24-07-1985
		CA 1255048 A	30-05-1989
		DE 3471942 A	14-07-1988
		US 4696979 A	29-09-1987
		CA 1254690 A	23-05-1989
		EP 0151883 A	21-08-1985
		SG 96994 G	12-05-1995
		US 4551501 A	05-11-1985
EP 0265075 A	27-04-1988	JP 1716143 C	27-11-1992
		JP 4002620 B	20-01-1992
		JP 63095252 A	26-04-1988
		CA 1300778 A	12-05-1992
		CN 1009103 B	08-08-1990
		DE 3774532 A	19-12-1991